

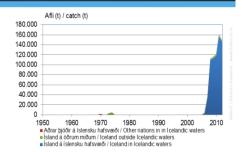
Atlantic Mackerel (Scombrus scomber)



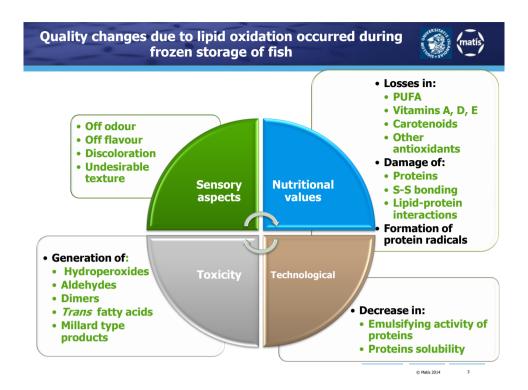


- Atlantic mackerel (Scombrus scomber) is one of the most valuable species in marine industry due to its nutritional value
- The mackerel is known from widespread relocations and has been discovered in Icelandic waters since 2006
- Icelandic mackerel is now exported mainly to Eastern Europe and gained great economical importance
- Frozen storage of the mackerel is main long term preservation method and it is a important task for industry to optimize this process in order to increase quality of the products





Onnur / other
Malæasia / 1.893 8%
Malaysia 449 2%
Portúgal / Portugal
757 3%
Litháen / Lithuania
1.764 7%
Nigeria / Nigeria
2.979 13%
Holland / Netherlands 4,184



Factors affecting lipid oxidation development during frozen storage





- 2) Geological variation of fishing grounds
- 3) Fish processing of raw material
- 4) Freezing methods
- **5)** Storage time (3, 6, 9, 12 months)
- 6) Storage temperature (-18°C, -25°C)

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Methods (matis)	
Water content	Weight loss during drying at 105°C for 4h (ISO 1983) / Near Infrared Reflectance measurements (NIR)
Lipid extraction	Bligh & Dyer method (1959)
Phospholipid content	Colorimetric method (Stewart 1980)
Fatty acids composition	Gas chromatography of fatty acid methyl esters (FAMEs) according to the AOCS (1998).
Free Fatty acids (FFA)	Lowry & Tinsley (1976), with modifications from Bernardez et al. (2005)
Thiobarbituric Acid Reactive Substances (TBARS)	A modified method of Lemon (1975)
Peroxide values (PV)	A modified method of the ferric thiocyanate method (Shantha & Decker 1994)
Fluorescence compounds	Perkin-Elmer LS 50B fluorescence spectrophotometer
Color measurement	DFK 31BF03 CCD digital camera. LensEye software

Effect of storage time and temperature on hydroperoxide rate





- Peroxide value (PV) primary oxidation products, main initial product of autoxidation
- PV significantly increased during the frozen storage of mackerel

Control
$$< 3m < 6m < 9m < 12m$$

The rate of formation hydroperoxide after 9 months and 12 months storage was significantly higher in the samples stored at -18°C comparing to the samples stored at -25°C

Control < -25°C < -18°C

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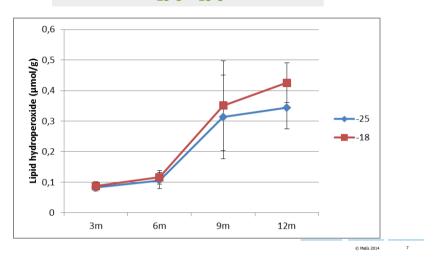
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Effect of storage time and temperature on lipid peroxide development





The rate of peroxide (PV) was significantly lower at -25°C at -18°C after 12 months storage -25°C < -18°C



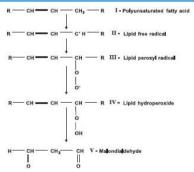
Effect of storage time and temperature on Thiobarbituric acid reactive substances (TBARS)

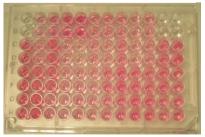




- TBARS secondary oxidation products arise mainly as a result of decomposition of primary oxidation products
- TBARS rate was rather stable during frozen storage but was significantly higher in the samples stored at -18°C comparing to the samples stored at -25°C

Control < -25°C < -18°C



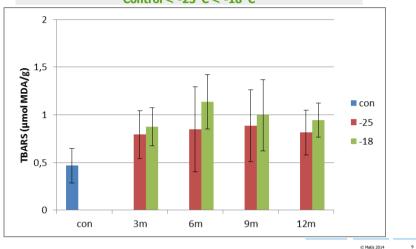


Effect of storage temperature on TBARS development









Effect of storage time and temperature on Free fatty acids (FFA)





- Free fatty acids (FFA) arise mainly in enzymatic hydrolysis of phospholipids during frozen storage. Furthermore, accumulation of FFA may leads to interaction with proteins and causes quality lost
- > FFA rate increased significantly with storage time

Control < 3m < 6m < 9m < 12m

➤ FFA were significantly higher in the samples stored at -18°C than in the samples stored at -25°C

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Effect of storage temperature on FFA development





FFA were significantly higher in the samples stored at -18°C than -25°C Control < -25°C < -18°C

